The chemical characteristics of composted and vermicomposted cotton residues case study in Sudan

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Abstract

Purpose - The purpose of this paper is to investigate the influence of two decomposition processes, namely, composting and vermicomposting, on the chemical composition of the finished products of a mixture of: cotton residues; soil and cotton residues; farmyard; soil.

Design/methodology/approach – Composting experiments were done over six months to prepare four different mixtures as follows: cotton residues + soil (C); cotton residues + soil + earthworms (C + E); cotton residues + soil + farmyard manure (C + F); and cotton residues + soil + farmyard manure + earthworms (C+F+E). Electrical conductivity, pH, nitrate-N, ammonium-N, ash, total phosphorus, total nitrogen, total organic carbon, carbon: nitrogen ratio, total potassium and trace elements (Mn, Fe, Cu and Zn) were determined on monthly-based samples.

Findings – Significant differences (b < 0.05) in organic carbon, nitrate-N, nitrogen, phosphorus, and potassium content were recorded in vermicompost compared to compost. In general, results indicated that vermicompost had a significant effect compared to compost and a positive effect on the chemical properties of the finished products.

Originality/value - This research work was carried out by four researchers from two institutions concerned with agricultural production and environmental aspects related to soil productivity. The paper emphasizes on production of organic fertilizers with good quality and monitoring of composting process for better management practices of agricultural wastes in Sudan.

Keywords Sudan, Soil mechanics, Cotton, Cotton residues, Farmyard manure, Composting, Vermicomposting

Paper type Case study



Introduction

Sudan is one of the two largest countries in Africa; it covers an area of about 1,881,000 km² after separation of South Sudan. The arable land constitutes about one-third of the total area of the country (Higher Council of Environment and Natural © Emerald Group Publishing Limited Resources, 2003).

World Journal of Science, Technology and Sustainable Development Vol. 9 No. 4, 2012 pp. 325-336 2042-5945 DOI 10.1108/20425941211271522 Applicable amounts of tons of cotton, wheat and other crops residues are burnt annually in irrigated schemes of Gezira-Managil (one of the largest state owned farms in the world), Rahad and Souki. These schemes, with heavy clay soils, extend over three million hectares. Cotton residues (CR) are in particular burnt in order to prevent the spread out of residue-borne diseases such as bacterial blight (black arm disease). This practice may lead to many environmental problems such as contamination of water, soil and plants since burning the crop residues together with the residual herbicides and insecticides produce toxic materials and gases. Furthermore, burning of such residues interferes with natural decomposition of organic matter in the soil and causes loss of a valuable source.

Crop residues are vital resource for the maintenance of soil productivity, since not only the residues are the primary substrate for the replenishment of soil organic matter, but also serve as an important source of plant nutrients (Janzen and Kucey, 1988). Retention of crop residues, when converted to organic matter, can improve soil physical, chemical and biological properties. Addition of the organic matter can improve soil chemical properties such as: increase the organic carbon (OC) content, cations exchange capacity (CEC) and consequently availability of plant nutrients (Giusquiani *et al.*, 1995).

Composting is defined as the decomposition or breakdown of organic waste by a mixed population of micro-organisms in a warm, moist, aerated environment (FAO, 1987) while decomposition under controlled conditions involving earthworms is called vermicompost (Viljoen *et al.*, 1992).

Material and methods

Experimental site

Compost and vermicompost were prepared over six months at the demonstration farm of the Faculty of Agriculture, University of Khartoum, Sudan (latitude 15°14′N and longitude 32°32′E and elevation of 380 meters above mean sea level).

Materials

CR were collected from Gazira Scheme, Gazira State, Sudan. The residues were first chopped manually and then mechanically using a heavy truck until an optimum size (less than two-inch length) was obtained. Farmyard manure (F) was collected from cow farm located in the Faculty of Veterinary Medicine, University of Khartoum, Sudan. Cows were normally fed with concentrate plus Abu Sabin (*Sorghum bicolor*). While, earthworms (V) (*Eisenia foetida*) were collected from commercial farms in Jabal Awlia and Toti Island Khartoum State, Sudan.

Methods

Composting and vermicomposting experiment

The decompostid mixture consisted of a mixture of soil, farmyard manure (F) and CR at 1:1:5 ratios on dry weight basis. Other mixtures were prepared without addition of farmyard manure which mixed at the ratio 1:5 (soil:CR). Compost (without earthworm) was made in pits $(2 \times 1 \times 0.7 \, \text{m})$ and vermicompost was prepared in $(2.4 \, \text{m}^3)$ plastic barrels. The rate of the inoculated earthworms was about $2 \, \text{kg/m}^2$ including eggs, larvae and adults according to Dominguez and Edwards (1997). Earthworms were added one month after the initiation of the composting process to avoid exposure of earthworm to high temperature during the initial stages of composting. Water was sprinkled every two days using an iron sprayer to maintain the moisture level of 60-80

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percent, barrels were kept under shade to avoid direct sunlight and high temperature Characteristics of that adversely affect earthworms. Measurement of temperature was taken every three days with digital thermometer. Composted and vermicomposted piles were turned every two weeks to get homogeneity. Samplings were taken monthly for chemical analysis till the end of the experiments that lasted for six months.

Samples analysis

Compost and vermicompost were analyzed at the Department of Soil and Environment Science, Faculty of Agriculture, University of Khartoum and Environmental and Natural Resources Research Institute (ENRRI), National Center for Research (NCR), Sudan.

The pH was measured in suspension of 1:5 (soil:CaCl₂) according to Hendersont et al. (2008) while the electrical conductivity (EC)_{1.5} was determined in suspension (soil:water). Available phosphorus (P) was measured calorimetrically by the molybdenum blue method (Jackson, 1958). Nitrogen (N) was estimated by micro-Kieldahl method (Association of Official Analytical Chemists, 1990) and OC was determined by Walkley and Black (1934) method. Dry ashing was conducted to prepare the samples for the determination of calcium (Ca²⁺), magnesium (Mg²⁺), sodium (Na⁺) and potassium (K⁺) according to Chapman and Pratt (1961). Sample extraction was performed following the procedure of Perkin (1994) for the determination of manganese (Mn), iron (Fe), copper (Cu) and zinc (Zn). Mineral nitrogen (NO₃-N and NH₄+-N) in compost and vermicompost were measured according to Bremner and Keeney (1965).

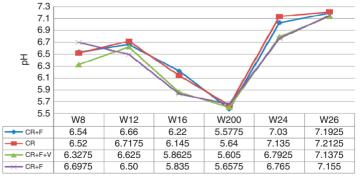
Statistical analyses

Analysis of variance was followed to test the significant differences among variables. Means were separated using Duncan's multiple range test. Unless otherwise stated, null hypothesis was rejected at p > 0.05. SAS software was used for all statistical analyses (SAS Institute, 2000).

Results and discussion

Effect of compost and vermin-compost on pH and EC

The pH of both treatments is depicted in Figure 1, the values increased at the beginning of the composting process to an alkaline neutral state. This increase was due



Notes: CR, cotton residues; F, Farmyard manure; V, earthworms; W, Week

Figure 1. Effect of treatments and time on pH during the periods of fermentation to the high-microbial activity and fast decomposition (Sundberg and Jönsson, 2008). The finished products from all piles showed a similar pattern of change in pH that was within optimal range for plant growth (5 and 6.5) (Goh and Haynes, 1977).

Figure 2 shows the change in EC during the decomposition, Results indicated that there were slight initial increases in all piles with time. However, for piles without earthworms a sharp increase was observed after the 12th week. The differences in EC values resulted from the addition of earthworms were significantly higher ($p \le 0.01$) over the whole period of decomposition. Atiyeh *et al.* (2000) reported that addition of earthworms could accelerate the decomposition and mineralization of organic matter. However, the EC value of both treatments was $<4\,\text{dS/m}$, which is considered appropriate for plant growth (Abid and Sayadi, 2006).

Effect of compost and vermicompost on ash content and C:N ratio

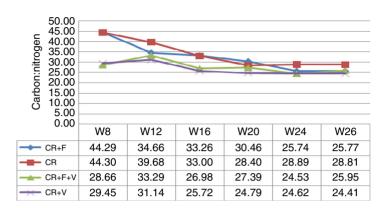
The ash content and C:N of both compost and vermicompost were shown in Figures 3 and 4, respectively. The results revealed that no significant difference (>0.05) in the ash content were observed between the two types of vermicompost (with and without farmyard manure). However, there was higher significant difference ($p \le 0.01$) between compost and vermicompost. The highest values of ash content were those of the compost during the whole period of decomposition. This result could be attributed to the rapid breakdown of C components and mineralization by the



Figure 2.
Effect of treatments and electrical conductivity during the period of fermentation

	90.00						
	85.00						_
	80.00						\triangleleft
%	75.00			4			
Ash%	70.00						
	65.00						
	60.00		-				
	55.00	W8	W12	W16	W20	W24	W26
_	CR+F	72.56	71.26	73.20	77.98	79.56	81.86
_	UN+F		71.20	73.20			
-	CR	75.81	75.01	77.82	80.45	83.30	85.19
<u>*</u>	CR+F+V	63.36	62.17	71.47	72.59	79.26	79.85
-	CR+V	63.34	61.76	74.91	73.63	80.39	78.25

Figure 3. Effect of treatments on ash content during the period of fermentation



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Figure 4. Effect of treatments and time on carbon:nitrogen during the period of fermentation

earthworms (Atiyeh *et al.*, 2000). The values of C:N of the four types of compost, were decreasing throughout the experiment period in four types of compost were decreasing throughout the experiment periods. The C:N ratios of vermicompost were lower over the whole period of decomposition. However, the final values of C:N ratio of all piles were between 24:1 and 29:1. On the other hand, it was observed that the C:N ratio in vermicompost initially increased, illustrating that much more rapid decomposition and rates of mineralization have been taken place. This result was in accordance with that reported by Vinceslas-Akpa and Loquet (1994). The gradual decrease in C:N ratio with time may be explained by the loss of OC as CO₂ (Gaur and Singh, 1995).

Effect of compost and vermicompost on nitrate-N (NO_3 -N) and ammonium-N (NH_4 -N) The concentration of NO_3 -N and NH_4 -N of both compost and vermicompost were shown in Figures 5 and 6. The concentration of NO_3 -N gradually decreased during the first 16 weeks of the decomposition and then increased to the higher values at the end of the decomposition period. The piles with CR plus farmyard manure (CR + F) showed the highest NO_3 -N values while the piles without farmyard manure (CR) showed the lowest concentration. NH_4 -N dropped from the initial values in the fourth week to lowest values in 24th week. The reduction was as follows: 69.67-43.62, 51.94-40.83, 68.54-55.02 and 61.45-57.85 mg/kg for CR + FYM, CR, CR + V and CR + FYM + V, respectively. The high concentration of the NH_4 -N at the initial

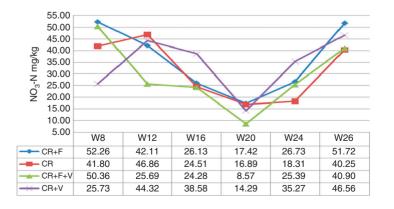
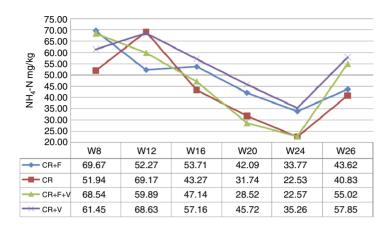


Figure 5.
Effect of treatments and time on the concentration of nitrate-nitrogen during the fermentation period

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Figure 6.
Effect of treatments and time on the concentration of ammonium-nitrogen during the fermentations period



stages of the decomposition might be due to the high temperature during thermophilic stage (from week four to week eight). Brady and Weil (2002), reported that when substrates temperatures during the decomposition exceeds 40°C, N generally volatilize as ammonia (NH₃⁺). Low concentrations of NH₄-N at the end of the decomposition period may be due to leaching caused by the water added to the compost (Phillips, 1988).

Changes in total organic carbon (TOC), total N and P during composting and vermicombosting

Table I shows the evolution of the TOC, total N and total P content during composting and vermicomposting. The TOC content decreased significantly in most of the composted mixtures throughout composting and vermicomposting time particularly after 24 weeks. Such a decrease TOC during the composting process can be considered as a clear indicator of the mineralization of the labile fractions of organic matter by the microorganisms present, which utilize these compounds as a C and energy source to build their own structure (Garcia-pichel and Castenholz, 1993). After 24 weeks, differences in total TOC were observed between different piles. The reduction in both composted and vermicomposted mixtures with farmyard manure (55 and 56 percent) are greater than that of mixture of no farmyard manure (41 and 53 percent), which can be attributed to the higher proportion of labile organic matter present in the farmyard manure (Hernandez *et al.*, 2006).

Concerning the total N content, results indicated significant ($p \le 0.01$) reduction of N in all mixtures. However, N in vermicompost was significantly ($p \le 0.01$) higher than in compost. This reduction was probably due to NH₃ volatilization and leaching into bedding material. Chaudhuril *et al.* (2000) reported that the relatively high level of N during the period vermicomposting is probably contributed by earthworms through excretion of NH₄⁺ and secretion of mucus. The final product had a lower concentration of N in all mixtures than the initial N. While the P had no significant differences (p > 0.05) among piles during the fourth and eighth week, but after the eighth week, the P increased significantly in all piles up to 16th week followed by slight drop through 20th and the 24th week. Similar results were obtained by Hamad (2009).

Changes in K^+ and trace elements during composting and vermicomposting Tables II and III shows the concentration of K^+ , Mg^{+2} , iron (Fe), Zn and Cu during the composting and vermicomposting. In all piles K^+ tends to increase gradually from

	w24	29.8^{ij}				
	w20	27.7^{J}				
P (ppm)	w16	37.2^{fgh}				(C)
Ь	w1	37.9^{fg}	$40.1^{\rm ef}$	57.5^{apc}	62.2^{a}	T) +00+ +0
	8w	13.1^{kl}	19.7^{k}	30.2^{ij}	41.3^{ef}	; ; ;
	w4	12.5^{1}				1000
	24	0.28^{g}	0.32^{fg}	$0.42^{\rm cdef}$	0.42^{cdef}	
	20	0.28^{g}	$0.39^{ m def}$	0.46^{cde}	0.46^{cde}	(E)
% NI	16	_	_	0.46^{cde}	_	(4)
Τ	12	0.5^{cd}				4:40
	8	$0.4^{ m def}$	0.5^{cd}	0.84^{ab}	0.88^{a}	G 2022 ±12.
	4	0.31^{fg}	0.41^{def}	0.79^{ab}	0.87^{a}	
	24	8.1^{jk}	8.2^{jk}	11.0^{hi}	10.5^{1}	+0m(c)
	20	9.9^{jk}				104
% JOL	16	7.3^{k}	11.2^{hi}	$10.7^{\rm hij}$	12.9^{ghi}	14th 41:45.
TOC	12	$15.6^{ m efg}$				1
	8	13.7^{1} 15.6^{kl}	$16.9^{\rm ef}$	25.8^{ab}	27.4^{a}	9
	4	13.7^{1}	18.2^{de}	23.7^{c}	23.8^{bc}	4
	Time (weeks) 4	C	C + F	C + V	C + F + V	Notes Money in the come column with different latings) are visualizable different (A / 00) and along the manificant tool (CD)

Notes: Means in the same column with different letter(s) are significantly different $(\rho \leq 0.05)$ according to least significant test (LSD)

Table I.
Changes in total
organic carbon, total
nitrogen and
phosphorus during
composting and
vermicomposting

Table II. Changes in potassium, magnesium and iron during composting and vermicomposting

Time (weeks)	4	∞	K + (ppm) 12 16	ppm) 16	20	24	24 4	~	Mg (p)	Mg (ppm) 12 16	20	24	w4	8w	Fe (Fe (ppm) w8 w12 w16	w20 w24	w24
C C + F C + F C + F + V	$1.3^{\rm bc}$ $1.07^{\rm c}$ $2.27^{\rm a}$ $2.27^{\rm a}$	1.3bc 1.08e 1.07c 1.27bc 2.27a 2.00abc 2.27a 2.11ab	0.96 ^{bcde} 1.18 ^{bcd} 2.13 ^{ab} 2.11 ^{ab}	$0.8^{\rm f}$ $1.01^{\rm e}$ $2.00^{\rm ab}$ $2.29^{\rm a}$	0.9 ^f 1.02 ^{ef} 1.79 ^{abcd} 1.87 ^{abc}	0.87 ^f 1.00 ^{ef} 1.35 ^{bc} 1.45 ^{bc}	0.69 ^{abcd} (0.76 ^{ab} (0.68 ^{abcd} (0.74 ^{abc} ().69abcd).75abc).66cd).74abc	0.68abcd 0.74abc 0.68abcd 0.67bcd	0.71 ^{abc} 0.73 ^{abc} 0.68 ^{abcd} 0.64 ^d	0.67bcd 0.68bcd 42.2fg 4fg 0.77abc 0.72abc 44.4gth 4fg 1 0.67cd 0.66cd 44.5g 4fg 0.63de 0.56e 45.9fg 4fg	0.68 ^{bcd} 0.72 ^{abc} 0.66 ^{cd} 0.56 ^e	42.2 ^{fg} 44.4 ^{gh} 44.5 ^g 45.9 ^{fg}	45.8 ^{fg} 44.9 ^{gh} 45.8 ^{fg} 46.8 ^{ef} 49.2 ^{bc} 48 ^{cd} 549.7 ^{bc} 50.4 ^b 5	44.9gh 46.8 ^{ef} 48 ^{cd} 50.4 ^b	7.8 ^{de} 8.7 ^{bcd} 0 ^b 3.5 ^a	42.4 ^{hi} 39.0 ^j 46.7 ^{fg} 47.5 ^{def}	36 ^j 38.7 ^j 46.45 ^{fg} 46.45 ^{fg}
Notes: Means in the same column with different letter(s) are significantly different ($\rho \le 0.05$) according to least significant test (LSD)	s in the	same c	olumn w	⁄ith diff	erent lett	er(s) are	e signific	antly dif	ferent (ϕ	≤0.05) a	ccording	to least	signifi	cant te	st (LSD	•		

			1177	ari (ppiri)						piii)		
Time (weeks)	4	8	12	16	20	24	4	8	12	16	20	24
C	0.087^{ab}	0.087^{ap}	0.77^{abcd}	$0.08^{ m abc}$	$0.08^{ m abc}$	0.070^{cd}	0.059^{a}	0.055^{a}	0.041^{bc}	0.44^{b}	0.038^{bc}	0.036^{bc}
C+F	$0.08^{ m abc}$	$0.09^{ m ap}$	$0.08^{ m apc}$	$0.08^{ m apc}$	$0.077^{ m apcd}$	0.070^{d}	0.055^{a}	0.053^{a}	$0.04^{ m bc}$	$0.042^{\rm b}$	$0.038^{ m bc}$	$0.036^{ m bc}$
C + V	$0.08^{ m abc}$	$0.086^{ m ap}$	0.071^{cd}	$0.064^{ m d}$	$0.064^{ m d}$	0.073^{cd}	$0.041^{ m bc}$	0.056^{a}	$0.041^{ m bc}$	0.035^{c}	$0.043^{\rm b}$	$0.041^{ m bc}$
C + F + V	0.088^{ap}	$0.086^{ m ap}$	0.071^{cd}	0.076	$0.076^{ m pcd}$	0.073^{cd}	0.038^{bc}	0.038^{pc}	0.038^{bc}	0.40^{bc}	0.40^{bc}	0.40^{bc}
Notes: Means in the same	in the same	column with	different	letter(s) are significantly	ď	ifferent ($p \leqslant 0.05$) according to leas	.05) accordi	ng to least	significant t	est (LSD)		

Table III.
Change in zinc and copper
during the composting
and vermicoposting

the fourth week up to the 12th week. The highest values were obtained in the vermicompost. On the other hand, a significant decrease was observed during the period of week eighth and week 16th in compost. However, until the 24th week no significant difference was observed among piles. Sharma (2003) reported that, in vermicomposting, the presence of microflora in the gut of earthworm might play an important role in K^+ content.

The concentration of the trace elements, namely, Mn, Fe, Cu and Zn of the finished products, was reduced in all piles. However, the pattern of change in these elements throughout the decomposition was not consistent. These results partially agree with that of Gratelly *et al.* (1996) who reported a reduction in total Cu concentration, but increase in total K, Ca, Mg, Fe, Mn and Zn concentrations during vermicomposting of dairy sludge.

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